

Shasha Hu

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with bases make stronger acids acids and bases interact with
each other in what is called a neutralization reaction

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increases the concentration of hydrogen ion H^+ in aqueous
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variety of common household substances and classify them
as acid base or neutral tests on shampoo produced the
following results felt slippery ph 8.5 turned red litmus blue
these results indicate shampoo is an acid shampoo is a base
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to determine if an unknown substance is an acid or a base
state the chemical definitions of an acid and a base in terms
of their behavior in water write the formula of the salt
formed when a given acid and base are combined

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and bases are weak that is they do not ionize fully in aqueous
solution a solution of a weak acid in water is a mixture of the
nonionized acid hydronium ion and the conjugate base of the
acid with the nonionized acid present in the greatest
concentration

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web oct 10 2022 correct answer d salt water explanation
neutralization is a chemical reaction between an acid and a
base resulting in the formation of a salt and water the
product of neutralization is therefore salt water as the acid
and base combine to form a salt compound and water
molecules rate this question 4